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NEWPORT RI 02841-1708

IN REPLY REFER TO:

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NON-CHROMATE CONVERSION COATINGS

TO WHOM IT MAY CONCERN:

BE IT KNOWN THAT (1) WAYNE C. TUCKER, and (2) MARIA G.

MEDEIROS, employees of the United States Government, citizens of
the United States of America, and (3) RICHARD BROWN, citizen of
the United Kingdom and residents of (1) Exeter, County of
Washington, State of Rhode Island, (2) Bristol, County of
Bristol, State of Rhode Island, (3) Wakefield, Washington County,
Rhode Island have invented certain new and useful improvements
entitled as set forth above of which the following is a
specification:

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1	Attorney Docket NO. 82602
2	
3	STATEMENT OF GOVERNMENT INTEREST
4	The invention described herein may be manufactured and used
5	by or for the Government of the United States of America for
6	Governmental purposes without the payment of any royalty thereon
7	or therefor.
8	
9	CROSS REFERENCE TO OTHER PATENT APPLICATIONS
10	This patent application is co-pending with one related
11	patent applications entitled NON-CHROMATE METAL SURFACE ETCHING
12	SOLUTIONS (Attorney Docket No. 82602), by the same inventors as
13	this application.
14	
15	BACKGROUND OF THE INVENTION
16	(1) FIELD OF THE INVENTION
17	The present invention relates to a non-chromate conversion
18	coating and method of treating a metal surface with same, and
19	more particularly, to a "drop-in replacement," such as a
20	titanate, for a chromate in a conventional conversion coating
21	solution that otherwise would contain the chromate.
22	(2) DESCRIPTION OF THE PRIOR ART
23	It is known that solutions containing hexavalent chromium
24	can be used to treat the surface of a metal, such as aluminum, t
25	effectively keep the metal surface from rusting. However,

- 1 although hexavalent chromium is an efficient rust-proofing agent,
- 2 it is highly toxic and adversely affects the environment and
- 3 human health. For this reason, many chromate-free chemical
- 4 conversion coatings for metal surfaces have been proposed.
- 5 Thus, various non-chromate conversion coatings, such as the
- 6 conversion coatings described in Tomlinson U.S. Patent No.
- 7 5,759,244, the disclosure of which is incorporated by reference
- 8 herein, have been disclosed which are designed to render a metal
- 9 less reactive in a corrosive environment. Such non-reactive or
- 10 less reactive metal surfaces produce a corrosion resistant outer
- 11 layer on the base metal or its oxide thereby leaving the
- 12 underlying metal protected from the environment. These coatings
- 13 are applied in one or more stages and are subsequently rinsed
- 14 with water to remove undesirable contaminants.
- 15 Chromate-free conversion coatings are therefore generally
- 16 based on chemical mixtures that react with a metal substrate
- 17 surface to form a protective layer. Many of these conversion
- 18 coatings are based on Group IV-B metals such as titanium,
- 19 zirconium and hafnium. For example, U.S. Patent No. 5,743,971 to
- 20 Inoue et al discloses a rust proof film-forming composition for
- 21 treating a metal surface comprising an oxidated substance, a
- 22 silicate and/or silicone dioxide and at least one member selected
- 23 from the group consisting of metal cations of titanium,
- 24 zirconium, cerium, strontium, vanadium, tungsten, and molybdenum.
- 25 A metal substrate is provided a rust proof film by immersing it

- 1 in the foregoing liquid rust proof film-forming composition.
- 2 Similarly, U.S. Patent No. 5,855,695 to McMillen et al discloses
- 3 a non-chrome passivating composition employed as a post-rinse for
- 4 enhancing the corrosion resistance of phosphated metal
- 5 substrates. The composition comprises the reaction product of an
- 6 epoxy-functional material containing at least two epoxy groups
- 7 and an alkanolamine, or a mixture of alkanolamines. The non-
- 8 chrome passivating composition further comprises a Group IV-B
- 9 metal ion, or a mixture of Group IV-B metal ions. Moreover, U.S.
- 10 Patent No. 5,897,716 to Reghi et al discloses a chemically and
- 11 thermally stable chromate-free aqueous liquid treatment for
- 12 metals for imparting corrosion resistance thereto. The chromate-
- 13 free aqueous liquid comprises components selected from the group
- 14 consisting of H₂TiF₆, H₂ZrF₆, H₂HfF₆, H₂SiF₆, H₂GeF₆, H₂SnF₆, HBF₄
- 15 and mixtures thereof.
- The shortcoming of conventional non-chromate conversion
- 17 coatings, such as those described above, is that they cannot be
- 18 integrated into and employed in place of chromates in current
- 19 metal treatment coatings which employ chromates. As such,
- 20 conventional non-chromate conversion coatings are usually
- 21 sufficiently different from previously employed chromate-
- 22 containing conversion coatings that significant changes are
- 23 required to be made in the metal treating process and in the
- 24 production of the conversion coating itself. These changes can
- 25 amount to substantial expenditures and usually require additional

- 1 approvals from the Department of the Navy or a regulatory agency
- 2 of the United States Government. Thus, there is a need for a
- 3 "drop-in replacement" that can be employed in place of chromate
- 4 compounds, such as sodium dichromate, now used in conventional
- 5 chromate conversion coatings. "Drop-in replacement" refers to a
- 6 compound that can be employed in a conventional conversion
- 7 coating in lieu or in place of a chromate without requiring any
- 8 or substantial changes in the make-up of the conversion coating
- 9 or its substituents.

10

11

SUMMARY OF THE INVENTION

- It is a primary object of the present invention to provide a
- 13 non-chromate conversion coating for treating metals which
- 14 contains a titanate in place of a chromate.
- It is a further primary object of the invention to provide a
- 16 "drop-in replacement" for a chromate that can be employed in a
- 17 conversion coating which otherwise would employ a chromate.
- 18 It is a further primary object of the invention to provide a
- 19 method of rust-proofing a metal substrate by applying a non-
- 20 chromate titanate conversion coating thereto.
- 21 Another object of the invention is to provide a one-stage
- 22 method of rust proofing a metal substrate by applying a non-
- 23 chromate titanate conversion coating thereto including sodium
- 24 metatinate and/or potassium titanate.

- 1 Another object of the invention is to provide a non-chromate
- 2 conversion coating that excludes therein organic additives,
- 3 structural component additives or chelating agents.
- 4 The objects of the invention are accomplished by providing a
- 5 highly effective, non-chromate conversion coating which includes
- 6 a titanate, such as sodium metatitanate or potassium titanate, in
- 7 lieu of a chromate in a typical conversion coating that otherwise
- 8 would contain a chromate.
- 9 The present invention is developed on the basis of findings
- 10 that an excellent rust proof film can be obtained by immersing a
- 11 metal substrate in an aqueous solution which includes sodium
- 12 dichromate, sodium fluoride, potassium ferricyanide and nitric
- 13 acid in an amount to provide a pH of 1.2 to 2.2. It is believed
- 14 that the chromate provides corrosion protection by way of a
- 15 cathodic reaction, specifically, the reduction of oxygen in the
- 16 presence of water:

$$O_2 + 2H_2O + 4e^- 4OH^-$$
 (1)

- 18 This cathodic reaction is similar for many systems, and by
- 19 changing the oxygen concentration in the solution, reveals the
- 20 cathodic behavior of the chromate. Moreover, when the reduction
- 21 of oxygen is the rate controlling reaction and chromates are
- 22 present, other metals and lower oxygen levels show similar
- 23 behavior, that is a lower or decreased limiting current density.
- 24 Test results show that a metal tested without a conversion
- 25 coating has a high limiting cathodic density. For example,

- 1 untreated Al2024T3 has a limiting cathodic current density of 10-
- 2 20 A/cm², however, when a chromate conversion coating is applied,
- 3 the cathodic limiting current density is lowered to $3-7 \text{ A/cm}^2$.
- 4 However, since personal exposure limits (PEL) for chromates is
- 5 0.1 mg/m³ (milligram per cubic meter), chromate containing
- 6 conversion coatings are not practical for use. Thus, a "drop-in
- 7 replacement" for the chromate in the chromate-containing
- 8 conversion coating is highly desired.
- 9 Sodium metatitanate and potassium titanate have been found
- 10 to be well suited as "drop-in replacements" for chromates in
- 11 conversion coatings which, in addition to sodium dichromate,
- 12 contain sodium fluoride, potassium ferricyanide and nitric acid.
- 13 For example, test results show that a conversion coating which
- 14 includes a "drop-in replacement" according to this invention in
- 15 place of a chromate produces a metal surface having a cathodic
- 16 limiting current density of 0.5 to 1 A/cm³. Furthermore, the PEL
- 17 for such a conversion coating is 15 mg/m³. Thus, the present
- invention provides a highly effective, non-toxic conversion
- 19 coating which otherwise would include toxic chromate compounds,
- 20 such as sodium dichromate.

21

- DESCRIPTION OF THE PREFERRED EMBODIMENT
- The present invention will hereafter be described in detail
- 24 with reference to the following embodiments.

- 1 The preferred embodiment of the present invention comprises
- 2 a non-chromate conversion coating for providing corrosion
- 3 resistance to metals which includes sodium metatitanate and/or
- 4 potassium titanate in a conversion coating that otherwise would
- 5 contain a chromate. The preferred embodiment further includes a
- 6 method of rust proofing a metal which employs the non-chromate
- 7 conversion coating of the present invention.
- 8 A typical chromate-containing conversion coating employed in
- 9 a metal treatment process includes sodium dichromate, sodium
- 10 fluoride, potassium ferricyanide and nitric acid, and more
- 11 particularly, about 0.025 M (molecular weight in grams per one
- 12 liter of solvent) sodium dichromate, about 0.024 M sodium
- 13 fluoride, about 0.015 M potassium ferricyanide and an amount of
- 14 nitric acid to provide a pH of 1.2 to 2.2. The present invention
- 15 provides a means of replacing this toxic metal treating solution
- 16 with a similar, non-toxic variant that includes the original non-
- 17 chromate constituents and thus, can be easily substituted for the
- 18 chromate-containing solution and employed in the same metal
- 19 treating process. Therefore, the preferred embodiment of the
- 20 present invention provides a conversion coating comprising sodium
- 21 metatitanate, sodium fluoride, potassium ferricyanide and an
- 22 amount of nitric acid to provide a pH of about 1.0 to about 6.0.
- 23 More particularly, the conversion coating of the present
- 24 invention comprises a solution of about 0-1 M sodium
- 25 metatitanate, about 0-1 M sodium fluoride, about 0-1 M potassium

- 1 ferricyanide and a balance of nitric acid to adjust the pH to
- 2 about 1.0 to about 6.0. Alternatively, potassium titanate can be
- 3 employed in place of sodium metatitanate. In that case, the
- 4 conversion coating comprises potassium titanate in an amount
- 5 ranging from about 4 g/l (grams per liter) to about 8 g/l, sodium
- 6 fluoride in an amount ranging from about 2 g/l to about 6 g/l and
- 7 nitric acid to adjust the pH to a range of about 1.0 to about
- 8 6.0.
- 9 Since the conversion coatings of the present invention are
- 10 drop-in-replacement compositions, additional additives, including
- 11 organic additives, structural component additives or chelating
- 12 agents for keeping the metals therein in solution are not needed.
- 13 Preferably, therefore, no such additives are included in the
- 14 compositions.
- To provide corrosive resistance to a metal surface by way of
- 16 the foregoing conversion coatings, the metal surface must first
- 17 be washed with a solvent, such as methanol or TCE
- 18 (trichloroethylene) in order to solvent wipe. Thereafter, the
- 19 surface is degreased with a 2% sodium hydroxide solution or any
- 20 other suitable degreaser such as a caustic solution for about one
- 21 minute, at about 50-60° C. Next, the metal surface is rinsed
- 22 with deionized water to remove any degreaser or solvent on the
- 23 metal's surface before being immersed in a deoxidizing solution
- 24 such as SMUTGO®. The metal surface is immersed therein for ten
- 25 minutes at about room temperature thereby deoxidizing the metal's

- 1 surface. Thereafter, the metal surface is again rinsed with
- 2 deionized water to remove any deoxidizing solution on its surface
- 3 before the non-chromate conversion coating of the present
- 4 invention is applied. It is preferred that the conversion
- 5 coating be around about 60-80° C during application. Lastly, the
- 6 metal is rinsed in a deionized water and allowed to air dry. An
- 7 advantage of the present method is that the non-chromate
- 8 conversion coating herein has only to be applied once to the
- 9 metal substrate, thus, the present method is a one-stage process.
- 10 Prior art non-chromate coatings and methods of applying same can
- 11 require multiple applications. Further, by rinsing the metal
- 12 surface following applying the present conversion coating, a dry-
- on polymer surface coating is not disposed on the metal surface
- 14 as is the case with prior art coatings.
- 15 A metal substrate, such as aluminum, that undergoes the
- 16 foregoing treatment is provided a lower cathodic limiting current
- 17 density than if allowed to go untreated. Specifically, test
- 18 results show that application of the non-chromate conversion
- 19 coating of the present invention to Al2024T3 results in a
- 20 cathodic limiting current density of 0.5 to 1 A/cm². Test
- 21 results were attained using a salt spray test over ten days.
- 22 While the preferred embodiment of the non-chromate
- 23 conversion coating and method of applying same has been described
- 24 in detail above, various modifications and variations of the
- 25 invention are possible in light of the above teachings. For

- 1 example, boric acid can be employed in place of nitric acid to
- 2 adjust the pH of the conversion coating. It is therefore
- 3 understood that within the scope of the appended claims the
- 4 invention may be practiced otherwise than above-described.

1	Attorney Docket No. 82602
2	
3	NON-CHROMATE CONVERSION COATINGS
4	
5	ABSTRACT OF THE DISCLOSURE
6	A non-chromate conversion coating and method of
7	applying same wherein the coating comprises a titanate, such as
8	potassium titanate or sodium metatitanate, as a "drop-in
9	replacement" for a chromate in an otherwise chromate-containing

conversion coating.